INDIAN RUBBER INSTITUTE PGD-IRI EXAMINATION – 2015

Paper - III

pate: 18th July, 2015

noration: 3 Hours

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+5) = 20

+4) = 20

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4) = 20

5) = 20

20

Time: 10.00-13.00 hrs.

Full Marks: 100

RUBBER MATERIALS

Answer should be illustrated with sketches wherever helpful

Total Five questions are to be answered. Each question carries 20 marks.

Ouestion No. 1 is compulsory. Answer Four questions from the remaining taking Two from each group.

GROUP - A

Multiple choice Questions: Select the correct answer from the given alternatives.

- i) Which rubber has the widest temperature range of application?
 - a) Poly Sulphide
 - b) EPDM
 - c) Chloro Sulphonated Poly ethylene (CSPE)
 - d) Silicone
- ii) Increasing acrylonitrile content, the property which decreases is:
 - a) Heat resistance
 - b) Heat Build Up
 - c) Glass transition temperature
 - d) Resilience
- iii) Good stabilizing agent for NR Latex
 - (a) Formic acid (b) Ammonia (c) Hydrochloric acid (d) Calcium Carbonate
- iv) Rubber which shows best gum strength
 - a) EPDM (b) BR (c) NR (d) SBR
- v) The basis for grading ISNR is:
 - a) Dirt content (b) Ash content (c) Viscosity
- vi) Amongst following rubbers, which rubber can take maximum fillers and oil loading
 - a) EPDM
- b) NR
- c) PU
- d) SBR

(d) Nitrogen content

- vii) Paraffin mineral oil is the most compatible with
 - a) Butyl rubber
 - b) Poly Chloroprene Rubber
 - c) Natural rubber
 - d) Nitrile rubber

vi	ii) PF Resin (a) NR	curing is the m	ost suitable for (c) IIR	(d) NBR		
ix) Curing sys (a) EV	(b) conv	ex life of cured ventional	product (c) Semi EV	(d) Peroxide	
x)	The most v (a) Arami		tile for reinforce Polyester	ement of V-belt (c) Nylon	(d) Rayon	
xi)	A rubber c a) 25	ompound cont b) 50	ain 25% carbon c) 75 d) (and 50% rubber Cannot be calcul	: Calculate the phr of cated	earbon black
xii	In latex processing the chemical used as gelling agent is a) Ammonia b) ZnO c) Sodium silicofluoride d) ZDEC					
xiii) Which of the a) SBR	b) CR	has low Mass c) BR	per Volume d) EPDM		
xiv	The best filler for Acid Resistant Tank Lining (a) ZnO (b) BaSO ₄ (c) Calcium Silicate (d) China clay					
xv)	xv) Polymer suffers from cold Flow					
	(a) CR	(b) NR	(c) SBR	(d) NBR		
xvi)	The most de (a) MBT	layed action action action (b) MBTS	ccelerator is: (c) TBSI	(d) NOBS		
xvii) Peroxide curing is not technically recommended for:-						
	a) NR	b) IIR	c) CR	d) NBR		
xviii)	Polymer is ha a) HDPE	ving highest in b) BIIR	npermeability o	ofair d) IIR		
xix)	Rubber hot w	ater bottle is r	nade from:			
	(a) BR	(b) IIR	(c) SBR	(d) Silicone Ru	ibber	1
xx)	Polymer which (a) EPDM	h shows the bo (b) VMQ	est combination (c) HNBR	of heat and oil re (d) CR	sistance:	
2.	Explain with p	xplain with proper reason of the following (any Four)				
a. The most preferred polymer for microwave cured profile for car is EPDM although it is not a polymer.						th it is not a pole to pole
						[Turn Ovel 1000]

- b. Natural rubber needs to be masticated where as no mastication is required for SBR, NBR, Polybutadiene.
- c. In bridge bearing pad, poly-chloroprene rubber is widely used.
- d. In XLPE cable EVA is a preferred polymer in conductor and semi conductor compound
- e. In NBR based compound higher dosage of SRF type black and oil is used to improve volume swell in mineral oil
- Green Strength of NR is higher than synthetic Polyisoprene.

$$4 \times 5 = 20$$

- 3.(a) Describe briefly the manufacturing of both smoked sheet rubber and Technically Specified rubber.
- (b) Explain how both the above rubbers (RSS & TSR) are graded ?
- (c) Explain what is the significance of PRI test values?

$$10 + 7 + 3 = 20$$

4. i) Design a rubber compound based on natural which will have a hardness of 65 shore A, tensile strength of 23 MPa, elongation at break of 700% Calculate the density and cost (both per kg and per volume) of the compound.

(Given density of NR = 0.92, ZnO = 5.57, St.acid = 0.85, S = 2.07, carbon black = 1.80, Accelerator = 1.27 or 1.37 or 1.28, A.O = 1.14 or 1.2 or 1.0). (all densities are in gm/cc). Justify your choice.

- Which rubbers/rubber blends you will choose for the following applications:
 Give reasons for your selection.
 - a) Tyre tread compound with good skid resistance.
 - b) V-belt base compound for use in oil contaminated area.
 - c) Mining boot soles.
 - d) Radiator hose.
 - e) Seals and 'O' rings for space vehicles.
 - f) Dock fender
 - g) Cable sheath
 - h) Inner liner for hoses carrying corrosive chemicals
 - i) Rubberized roll in the paper industry
 - j) Engine mountings.

 $(4+3+3)+1\times10=20$

GROUP - B

- 5. (a) What is carbon black? Write briefly the manufacturing process of furnace carbon black.
 - (b) What are the important properties of carbon blacks? What is meant by "Structure" of carbon blacks?
 - (c) What is the ASTM Nomenclature of N-219 stands for ? Write a Table on the effect of changes in particles size and structure on rubber compound properties.

$$8 + 4 + 8 = 20$$

What do you understand by Recycling. State the importance of recycling of tire with respect to was control in manufacturing unit, preservation of environment & society and producing value added product.
 Outline the different technologies of recycling, of tires. Compare the different recycle material produced o of these technologies.

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- Recycle material can be used as one of significant material to be used as compounding ingredients for cos (c) reduction in product manufacturing - Justify.
 - (2+3x2)+(4+4)+4=20

8 + 4 + 4 + 4 = 20

- 7. (a) Write the most suitable elastomers for each of the following and give reasons why?
 - White sidewall for passernger car tyre
 - Inner tube for oil filled Hose (ii)
 - (iii) Flame Retardant cover compound for conveyor belt.
 - High voltage cable Insulation. (iv)
 - Tyre curing bag for Automobile Tyre. (v)
- What is Aniline point? How the value of Aniline point helps to assess the type & quality of a plasticizer? (b) (c)
- How the mechanical stability of latex is tested and what is its significance.
- Write briefly the comparison amont conventional, EV and semi EV Systems. (d)
- Write short notes on any five: 8.
 - Fluorocarbon Rubbers. (a)
 - (b) Polysulphide Rubber
 - (c) Tackifiers
 - Flame retardants and smoke depressant. (d)
 - (e) Solution SBR
 - Reinforcing white fillers (f)
 - (g) Ester Plasticizers
 - Epoxide Natural rubber (h)

 $5 \times 4 = 20$

4 = 20